

ERRATUM

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Erratum to: Anoxic metabolism and biochemical production in *Pseudomonas putida* F1 driven by a bioelectrochemical system

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Erratum to: *Biotechnol Biofuels* (2016) 9:39 DOI 10.1186/s13068-016-0452-y

After the publication of the article [1], it was brought to our attention that some of the data in Table 2 were incorrect. Please find a correct and updated version of Table 2 in the erratum. Following this Fig. 1 has also been updated; the correct version of Fig. 1 is given in this erratum.

Also, during the calculation of specific glucose uptake rate, the authors mistakenly used the unit mmol/L as mmol, and therefore it caused some errors in the calculations of production rate (Table 2) and ATP regeneration rate (Section “Flux balance analysis”—[1]) which need to be corrected. The corrected ATP regeneration rates are 0.02 and 0.38 mmol_{ATP}/(gCDW h) for [Co(bpy)₃](ClO₄)₂ from glucose oxidation and membrane-bound ATP synthase respectively, while those numbers for K₃[Fe(CN)₆] are 0.05 and 0.64 mmol_{ATP}/(gCDW h), respectively.

Table 2 Key process parameters of anaerobic glucose conversion of *P. putida* F1 in the anode compartment of a BES using [Co(bipy)₃]^{3+/2+} or [Fe(CN)₆]^{3-/4-} as electron acceptors while poisoning the anode at +0.697 V vs SHE

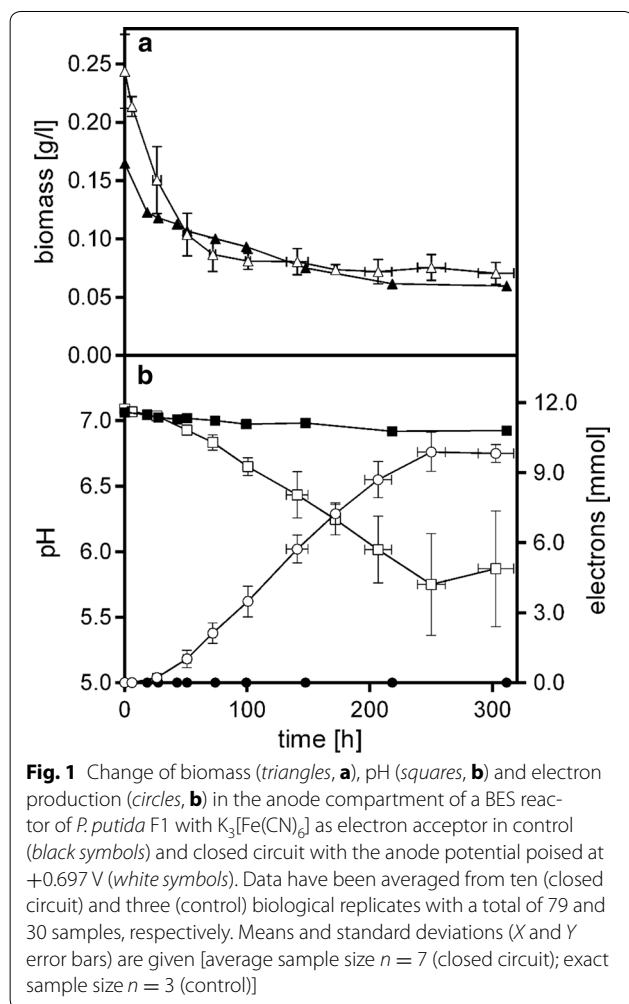
	[Co(bipy) ₃] ^{3+/2+}	[Fe(CN) ₆] ^{3-/4-}
Carbon balance (%)	99.6	97.6
Coulombic efficiency (%)	98.5	93.3
Yields (mol _{product} /mol _{glucose})		
Y _{2KGA}	0.90 ± 0.03	0.90 ± 0.02
Y _{acetic acid}	0.073 ± 0.008	0.144 ± 0.012
Y _{gluconic acid}	0.31 ± 0.06	0.09 ± 0.03
Y _{electrons}	0.25 ± 0.03	0.09 ± 0.04
Rates (mmol/(gCDW h))		
r _{glucose}	3.94 ± 0.11	3.88 ± 0.07
r _{glucose}	−0.26 ± 0.04	−0.35 ± 0.07
r _{acetic acid}	0.019 ± 0.003	0.051 ± 0.010
r _{2KGA}	0.23 ± 0.04	0.32 ± 0.06
r _{gluconic acid}	0.08 ± 0.02	0.03 ± 0.01
r _{gluconic acid}	−0.06 ± 0.01	−0.03 ± 0.02
r _{electrons}	1.02 ± 0.18	1.37 ± 0.26

Data are fitted with linear regression using datasets from ten ([Fe(CN)₆]^{3-/4-}) and four ([Co(bipy)₃]^{3+/2+}) biological replicates with a total of 79 and 36 samples, respectively (compare Additional file 1: Fig. S3). Carbon balance is calculated from the fitted rates considering carbon content of molecules and assuming equimolar CO₂ production when making acetate from glucose. Gluconic acid is a product in the first 100 h and serves as a substrate thereafter, hence 2 yields and rates are given

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